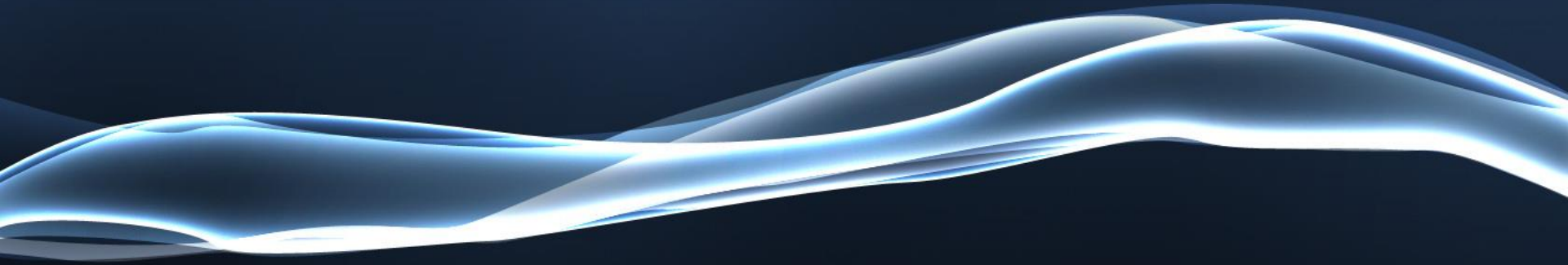


EXPONENTIAL GROWTH & DECAY



**The exponential function is
One of the most important functions
In the applications of mathematics...**

**E.G...it is used in the study of
Bacteria growth in a lab,
Decay of radioactive material,
And calculating interest earned
In a bank account...!**

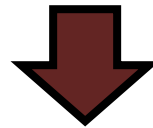
If y is a function of t , such that:

$$y(t) = ka^t$$

Where $k > 0$ & a are constants.

$a > 0$ & a cannot be 1

Then we say that y is growing/decaying exponentially.



Growing if: $a > 1$

Decaying if: $0 < a < 1$

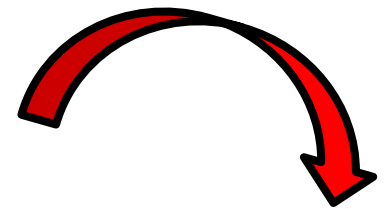
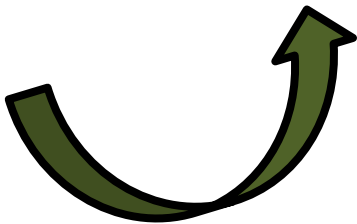


Can also be written as:

$$y(t) = ke^{at}$$

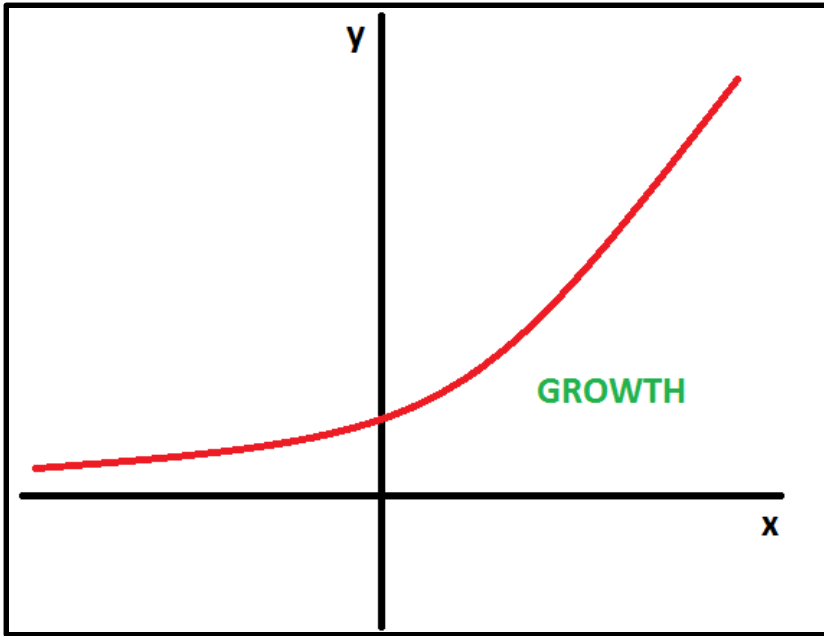
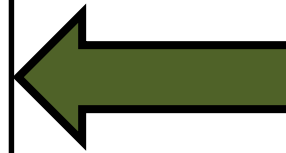
$a > 0$...**GROWING**

$a < 0$...**DECAYING**

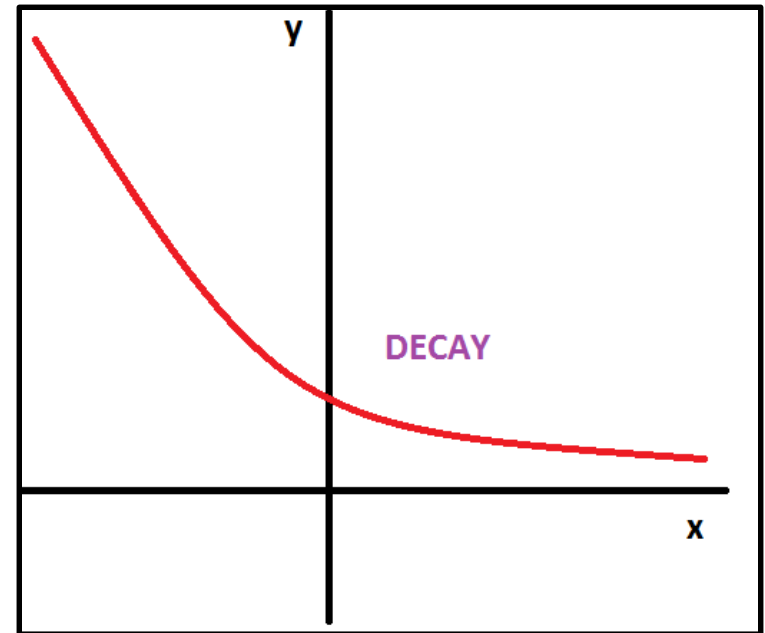
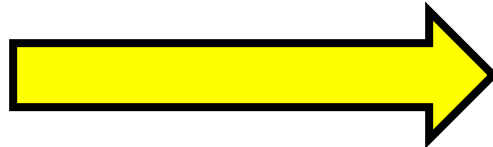




GROWTH



DECAY



Meaning of k & a

The initial value is the value of y at time zero.

$$y(0) \text{ when } t = 0$$

$$y_0 = ka^0 = k$$

$\therefore$ Constant k = initial value of y

POW!



$$y(t) = ka^t$$

$$y(t + 1) = ka^{t+1}$$

$$\frac{y(t + 1)}{y(t)} = \frac{ka^{t+1}}{ka^t} = a$$

$$\therefore y(t + 1) = ay(t)$$

Are you ready for a derivation...?

...YES!!

Are you sure...?



Well...check this out...



generally, suppose we know the value of the function at 2 different times ..

$$y_1 = ce^{kt_1} \dots (1)$$

$$y_2 = ce^{kt_2} \dots (2)$$

dividing the 1st equation by the second equation ...

$$\frac{y_1}{y_2} = e^{k(t_1 - t_2)}$$



taking natural log ...

$$\ln y_1 - \ln y_2 = k(t_1 - t_2)$$

dividing by $t_1 - t_2$...

$$k = \frac{\ln y_1 - \ln y_2}{(t_1 - t_2)}$$



substituting c back ...

$$y_1 = ce^{\frac{\ln y_1 - \ln y_2}{(t_1 - t_2)} \cdot t_1}$$

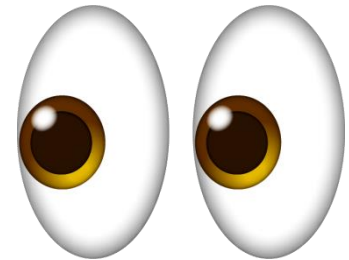
taking natural log ...

$$\ln y_1 = \ln c + \frac{\ln y_1 - \ln y_2}{(t_1 - t_2)} \cdot t_1$$

in summary we have 2 equations:

$$y_1 = ce^{kt_1} \dots (1)$$

$$y_2 = ce^{kt_2} \dots (2)$$



allows us to solve for c ...

$$\therefore c = e^{\left(\frac{t_1 \cdot \ln y_2 - t_2 \ln y_1}{t_1 - t_2}\right)}$$

SHALL WE SOLVE SOME EXAMPLES TOGETHER...

Example 1:

The mass of a colony of bacteria grows exponentially. Initially the colony has a mass of 2mg.

3 hours later...the mass is 2.1mg.

- a) Express the mass of the colony as a function of time.
- b) How long does it take for the mass to be 5mg?

let $y(t)$ be the mass of the colony
after t hours ...

$$y(0) = 2$$

$$y(3) = 2.1$$

$$y(t) = ka^t$$

$$k = 2$$

$$ka^3 = 2.1$$

$$\frac{2.1}{2} = a^3$$

$$a = \sqrt[3]{1.05}$$

$$a = 1.05^{\frac{1}{3}}$$

finally ...

$$y = ka^t$$

$$= 2 \left[(1.05)^{\frac{1}{3}} \right]^t$$

$$= 2(1.05)^{\frac{t}{3}}$$

*we are asked to find the time
when the mass is 5mg ...*

$$5 = 2(1.05)^{\frac{t}{3}}$$

$$\log \frac{5}{2} = \log (1.05)^{\frac{t}{3}}$$

$$\frac{t}{3} = \frac{\log 2.5}{\log 1.05}$$

$$t = 3 \times 18.8$$

$$t = 56.3 \text{ hours}$$

Example 2:

The amount of radioactivity given off by a substance decrease by 5% every hour.

After how long will it be 1% of its original value?



let the original value be R

After t hours, the amount will be $R(0.95)^t$

$$\frac{R}{100} = R(0.95)^t$$

$$\log \frac{1}{100} = t \log 0.95$$

$$t = -\frac{2}{\log 0.95}$$

$$t = 89.78 \text{ hours}$$



Example 3:

A herd of llamas has 1000 llamas in it.

The population is growing exponentially.

At time $t = 4$...it has 2000 llamas

Write a formula at any time t .



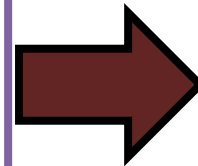
$$f(t) = c \cdot e^{kt}$$

where $f(t)$ is the number of llamas at time t .
 c and k are constants to be determined

at $t = 0$... there are 1000 llamas
 at $t = 4$... there are 2000 llamas

$c = \text{value of } f \text{ at } t = 0 \dots \text{which is } 1000$

$$\begin{aligned} k &= \left(\frac{t_1 \cdot \ln y_2 - t_2 \ln y_1}{t_1 - t_2} \right) \\ &= \frac{\ln 1000 - \ln 2000}{0 - 4} \\ &= \ln \frac{1000}{2000} - 4 \\ &= \frac{\ln 0.5}{-4} \\ &= \ln \frac{2}{4} \end{aligned}$$



$$f(t) = 1000e^{\frac{\ln 2}{4}t}$$

$$f(t) = 1000 \cdot 2^{\frac{t}{4}}$$



Example 4:

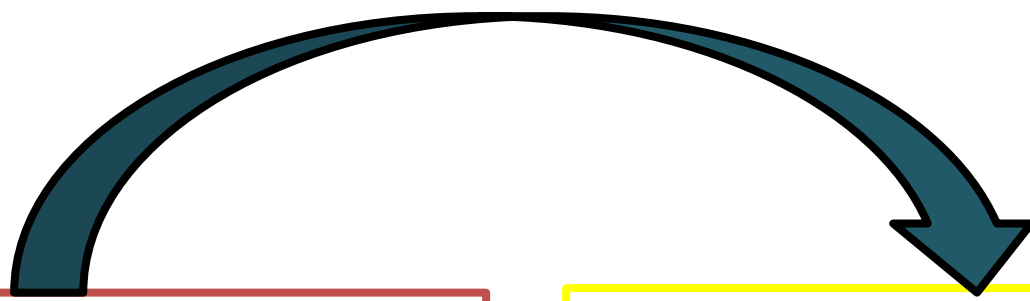
A colony of bacteria is growing exponentially.

At time $t = 0$...it has 10 bacteria

At time $t = 4$ it has 2000 bacteria

At what time will it have 100 000 bacteria?





$$k = \left(\frac{t_1 \cdot \ln y_2 - t_2 \ln y_1}{t_1 - t_2} \right)$$

$$= \frac{\ln 10 - \ln 2000}{0 - 4}$$

$$= \frac{\ln \frac{10}{2000}}{-4}$$

$$= \frac{\ln 200}{4}$$



$$\begin{aligned} f(t) &= 10 \cdot e^{\ln \frac{200}{4} \cdot t} \\ &= 10 \cdot 200^{\frac{t}{4}} \end{aligned}$$



$$100\,000 = 10 \cdot e^{\ln \frac{200}{4} \cdot t} \dots\dots \div 10$$

$$\ln 10\,000 = \ln \frac{200}{4} \cdot t$$

$$t = 4 \left(\frac{\ln 10000}{\ln 200} \right) \dots\dots \text{solving for } t$$

$$t = 6.95$$

The half life

Given the process of exponential decay...we can ask how long it would take for the half the original amount to remain...

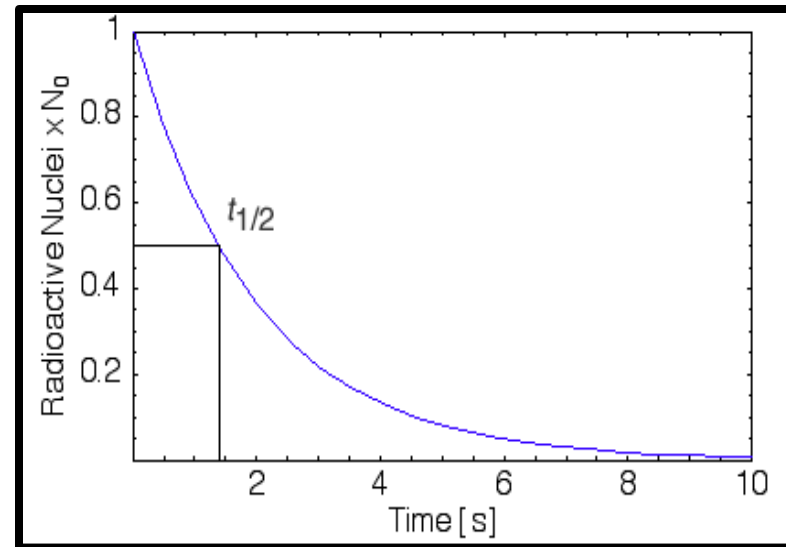
$$y(t) = \frac{y_0}{2}$$
$$\frac{y_0}{2} = y_0 \cdot e^{-kt}$$
$$\frac{1}{2} = e^{-kt}$$

taking reciprocals ...

$$2 = \frac{1}{e^{-kt}} = e^{kt}$$

$$\ln 2 = \ln e^{kt} = kt$$

$$\tau = \frac{\ln 2}{k}$$



*τ this symbol represents
half life ... !!!*



In 1986, the Chernobyl nuclear plant exploded.
Two radioactive elements were released.

These are **Iodine¹³¹** whose **half life is 8 days...**
& the other was **Cesium¹³⁷** whose **half life is 30 years.**

With our model for radioactive decay, we can
predict how much of this material would remain
over time...



Lets determine the decay constants

for iodine¹³¹

$$k = \frac{\ln 2}{\tau}$$
$$= \frac{\ln 2}{8}$$

$$k = 0.0866 \text{ per day}$$

*thus if t is measured in days ...
the amount of I^{131} would be ...*

$$y(t) = y_0 \cdot e^{-0.0866(t)}$$

for cesium¹³⁷

$$k = \frac{\ln 2}{\tau}$$
$$= \frac{\ln 2}{30}$$

$$k = 0.023 \text{ per year}$$

$$y(t) = y_0 e^{-0.023(t)}$$

***how long would I^{131} take to
decay to 0.1% of
its original value?***



$$0.001y_0 = y_0 \cdot e^{-0.0866(t)}$$

$$0.001 = e^{-0.0866(t)}$$

$$\ln 0.001 = -0.0866(t)$$

$$t = 79.7 \text{ days}$$